TVL10S

Tirunelvell District

Common Quarterly Examination - September 2025

| | | Standard | 10 | |
|-----|-------|---|--|------------------|
| Tim | ne: 3 | .00 Hrs. SCIENCE | | Marks: 75 |
| | | PART-I | | |
| I. | Cho | ose the correct answer: | | 12×1=12 |
| | 1) | In a myopic eye, the image of the obje | ect is formed. | |
| | , | a) behind the retina | b) on the retina | |
| | | c) infront of the retina | d) on the blind spo | ot |
| | 2) | The value of Universal gas constant | | |
| | | a) 3.81 J mol ⁻¹ K ⁻¹ | b) 8.03 J mol ⁻¹ K ⁻¹ | |
| | | c) 1.38 J mol ⁻¹ K ⁻¹ | d) 8.31 J mol ⁻¹ K ⁻ | 1 |
| | 3) | SI unit of resistance is | | |
| | 1.5 | a) mho b) Joule | c) ohm | d) ohm meter |
| | 4) | Which of the following is a triatomic mo | olecule?" | |
| | | a) Glucose b) Helium | c) Carbon di oxide | d) Hydrogen |
| | 5) | The basis of modern periodic law is | | |
| | | a) atomic number | b) atomic mass | • |
| | | c) isotopic mass | d) number of neutr | rons |
| | 6) | A solution is a mixture. | | |
| | | a) homogeneous | b) heterogeneous | gita Maria |
| | | c) homogeneous and heterogeneous | d) non homogeneo | us |
| | 7) | Which is formed during anaerobic respin | ration? | |
| | | a) Carbohydrate b) Ethyl alcohol | c) Acetyl CoA | d) Pyruvate |
| | 8) | 'Heart of Heart' is called | 11 21 42 10 20 20 20 10 10 10 10 10 10 10 10 10 10 10 10 10 | |
| | | a) SA node b) AV node | c) Purkinje fibres | d) Bundle of His |
| | 9) | Node of Ranvier is found in | | d) Cuton |
| | | a) Muscles b) Axons | c) Dendrites | d) Cyton |
| | 10) | LH is secreted by | | |
| | | a) Adrenal gland | b) Thyroid gland | |
| | | c) Anterior pituitary | d) Hypothalamus | |
| | 11) | Asexual reproduction takes place throu | gh budding in | d) Bryophyllum |
| | - 4 | a) Amoeba b) Yeast | c) Plasmodium | a) Bryophyllann |
| | 12) | Okasaki fragments are joined together | DY | d) DNA ligase |
| | 2 | a) Helicase b) DNA polymerase | c) KIVA Primer | u) DIVA ligase |
| | | PART-II | ulcore | 7×2=14 |
| п. | Ans | wer any seven questions. Q.No. 22 c | ompuisory: | |

- 13) State Newton's second law.
- 14) Distinguish between ideal gas and real gas.
- 15) Define the unit of current.
- 16) What is molar volume of a gas?
- 17) Give an example:
 - (i) gas in liquid (ii) solid in liquid (iii) solid in solid (iv) gas in gas
- 18) What is photosynthesis and where in a cell does it occur?
- 19) Define reflex arc.
- 20) Why are thyroid hormones referred as personality hormone? www.freetnbooks.com | FREETNBOOKS

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- 21) What do you understand by the term phenotype and genotype?
- 22) An object is placed at a distance 20 cm from a convex lens of focal length 10 cm. Find the image distance and nature of the image.

PART-III

III. Answer any seven. Q.No. 32 compulsory:

7×4=28

- 23) What are the types of inertia? Give an example for each type.
- 24) Differentiate the eye defects. Myopia and Hypermetropia.
- 25) i) Why is tungsten metal used in bulbs, but not in fuse wires?
 - ii) State ohm's law.
- 26) Derive the relationship between Relative molecular mass and vapour density.
- 27) In what way hygroscopic substances differ from deliquescent substances.
- 28) i) Write the dental formula of rabbit.
 - ii) How does locomotion take place in leech?
- 29) i) What is cohesion?
 - ii) Match the following:
 - a) Leukemia . . Phagocyte
 - b) Monocytes Pressure gradient
 - c) AB blood group. Blood cancer
 - d) Osmosis Absence of antibody
- 30) Write the physiological effects of gibberellins.
- 31) Classify neurons based on its structure.
- 32) Calculate the volume of ethanol in 200 ml solution of 20% v/v aqueous solution of ethanol.

PART-IV

IV. Answer all the questions:

3×7=21

33) a) State and prove the law of conservation of linear momentum.

(OR)

- b) With the help of a circuit diagram derive the formula for the resultant resistance of three resistances connected.
 - (a) in series and (b) in parallel
- 34) i) Give the salient features of "Modern atomic theory".
 - ii) What is rust? Give the equation for formation of rust.
 - iii) State two conditions necessary for rusting of iron.

(OR)

- i) Write notes on various factors affecting solubility.
- ii) What happens when MgSO₄.7H₂O is heated? Write the appropriate equation.
- 35) a) What are the phases of menstrual cycle? Indicate the changes in the ovary and uterus. (OR)
 - Explain with an example the inheritance of dihybrid cross. How is it different from monohybrid cross.

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